

## Density

Concentration of matter in an object.

Classroom with 35 students > density > classroom with 2 students

Density of object = ratio of mass to volume

Hence

$$\text{density} = \frac{\text{mass}}{\text{volume}} \quad \text{or} \quad d = \frac{m}{v} \quad \text{or} \quad d = m / v \quad * \text{see ESRT}$$

### Number must have units!

Volume = amount of space a substance occupies

Mass = amount of matter in an object

**e.g.:** mineral mass = 44 grams and volume = 20 cm<sup>3</sup>, what is the density?

$$d = m/v$$

$$d = \frac{44 \text{ g}}{20 \text{ cm}^3}$$

$$d = 2.2 \text{ g/cm}^3$$

**e.g.:** object's mass = 100 g and volume = 200 cm<sup>3</sup> calculate density.

$$\text{e.g.: } m = 10 \text{ g and } v = 5 \text{ cm}^3$$

**e.g.:** density = 5 g / cm<sup>3</sup> and mass = 20 g, what is the volume?

**e.g.:** d = 0.1 g / cm<sup>3</sup> and v = 50 cm<sup>3</sup> what is the mass?

Changes in temperature and pressure affect density.